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## Primary investigating chemical constituents of bioactive extract from *Centrostachys aquatica* (R.Br.) Wall. ex Moq.-Tand.

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### ABSTRACT

This study is aimed to screen the biological activities and chemical composition to find evidences for potential medicinal applications of *Centrostachys aquatica* in the Mekong Delta. Crude methanol extract and subextracts in *n*-hexane, ethyl acetate, and acetone from *Centrostachys aquatica* were tested bioactivities. The methanol extract, *n*-hexane and ethyl acetate subextracts exhibited antimicrobial activity with corresponding MIC values of 200, 100 and 200  $\mu\text{g/mL}$ , respectively. The ethyl acetate subextract was inhibited cytotoxicity against cancer cell line LU-1 with  $\text{IC}_{50}$  of 27.66  $\mu\text{g/mL}$ . None of the extracts showed antioxidant ability. Three known secondary metabolites including oleanolic acid (1), 20-hydroxyecdysone (2), and  $\beta$ -spinasterol (3) were isolated for the first time from the bioactive (ethyl acetate) subextract of *Centrostachys aquatica*. Their structures were elucidated by modern spectra as MS, NMR and comparison with published data.

### 1. INTRODUCTION

*Centrostachys aquatica* (R.Br.) Wall. ex Moq.-Tand. or *Achyranthes aquatica* R.Br. (Ho, 1999) has

a local name of “Cỏ Xước nước” in Vietnamese, it means “water scratch grass”. It is only one aquatic species belonging to the genus *Centrostachys* in the family *Amaranthaceae* (Figure 1).



**Figure 1. *Centrostachys aquatica* (R.Br.) Wall. ex Moq.-Tand.**

In Viet Nam, *Centrostachys aquatica* distributed in wetlands such as swamps, alluvial flats along rivers and canals in the Mekong Delta. People use this plant as a kind of vegetable and there is no information about its use in traditional medicine.

In recently researched documents on *Centrostachys aquatica* were found, there was only one report about isolating a compound called loliolide; the aqueous methanol extract and loliolide from this extract both have the ability to inhibit the growth of roots and hypocotyls of cress (*Lepidium sativum*) (Bich & Kato-Noguchi, 2014). This could be the basis for a convincing explanation for phytotoxicity and allelopathic activity of *Centrostachys aquatica*.

To find out if *Centrostachys aquatica* had other beneficial activities, extracts from this plant were tested some of the bioactivities. The results showed that the ethyl acetate extract has good antimicrobial and cytotoxic activity against lung cancer cells (section 3.1). This is also the reason for studying the chemical composition of this bioactive extract with three natural compounds were initially isolated.

## 2. EXPERIMENT

### 2.1. Plant material

The whole plants of *Centrostachys aquatica* were collected in Can Tho city in May, 2021. Voucher specimens have been identified at Department of Biology, School of Education, Can Tho University. After cleaning, poor quality parts were removed. Good material was dried at 50°C in order to decrease the humidity to less than 2%, followed by crushing into fine powder.

### 2.2. General experimental procedures

#### 2.2.1. Extraction and purification

Solid-liquid extractions were used with methanol, *n*-hexane, ethyl acetate and acetone. Solvent

evaporating was accomplished by using RE-52A rotary evaporator system (China).

Thin layer chromatography (TLC) was carried out on pre-coated silica gel 60F<sub>254</sub> (0.25 mm) aluminium sheet (Merck). Traces of compounds were detected by illuminating under UV light (254/365 nm) or spraying 10% H<sub>2</sub>SO<sub>4</sub> solution in ethanol and then heating at 105°C for 1-2 min on electric stove.

For common phase column chromatography (CP-CC), silica gel 60 (0.040-0.063 mm, Merck), increasing polarity solvent systems including *n*-hexane (H), chloroform (C), ethyl acetate (E) and methanol (M) was used. Compounds were purified by re-crystallization in pure solvents.

#### 2.2.2. Structural elucidation and identification

Melting point (mp.) was recorded by a melting point meter (Electrothermal 9100, UK), using capillary at Can Tho University. <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, DEPT, HSQC, COSY, HMBC spectra were recorded on a Bruker AM500, 600 FT-NMR spectrometer; Mass spectrum (MS) was recorded on mass spectrometer (HP 1100 series, LC/MSD Trap, Agilent) at Vietnam Academy of Science and Technology.

#### 2.2.3. Antimicrobial, cytotoxic and antioxidant activity testing

Biological activity assays were conducted at the Experimental Biology Department, Institute of Chemistry of Natural Compounds, Vietnam Academy of Science and Technology.

The antimicrobial activity assay was followed the disc diffusion agar method of Vanden-Berghe and Vlietinck (1991), Mckane and Kandel (1996), improved by using 96-well microplate and ELISA reader.

The cytotoxic activity assay was performed according to the method of Skehan et al. (1990) and Likhiwitayawuid and Angerhofer (1993) has been applied at the USA National Cancer Institute (NCI) and the College of Pharmacy, University of Illinois, Chicago, USA.

The antioxidant activity assay was based on its ability to trap free radicals generated by 1,1-diphenyl-2-picrylhydrazyl (DPPH) of Brand-Williams et al. (1995), Shela et al. (2003) and Kumar et al. (2013).

### 2.3. Extraction and isolation

The dried plant powder (7.0 kg) was exhaustively extracted with methanol 70%vol (> 20 L) and then evaporated under reduce pressure to remove the solvent to give dry dark-green residue of crude methanol extract (CAME, 305 g).

The CAME extract (300 g) was distributed with *n*-hexane (10 L), ethyl acetate (15 L), and acetone (10 L), respectively, and the solvents were evaporated under poor pressure to obtain subextracts CAHe (50.6 g), CAEt (93.7 g), CAAc (40.3 g) and the remainder was insoluble in distributed solvents (CAW, 101 g).

The CAEt subextract (90 g) was subjected to CP-CC with H:E (gradient, 0 to 100% E) and final with E:M (9:1) solvent mixtures as eluent to give 7 fractions (CAE1-7).

The fraction CAE2 (H:E 8:2; 13.12 g) was treated with CP-CC (H:E, gradient, 0 to 100% E) to afford 6 subfractions (CAE2.1-2.6). The subfraction CAE2.4 (H:E 85:15, 3.45 g) was continued to take CP-CC (C:M, gradient, 0 to 100% M) to get 5 subfractions (CAE2.4.1-2.4.5). The subfraction CAE2.4.3 (C:M 95:5, 0.55 g) was washed with H:E 9:1 and then re-crystallized in methanol to produce compound **1** (10.1 mg).

The fraction CAE5 (H:E 25:75, 23.52 g) was performed CP-CC with H:E solvent systems (from

5:5 to 25:75, ending with methanol) to afford 6 subfractions (CAE5.1-5.6). The subfraction CAE5.2 (H:E 5:5, 4.21 g) was cleaned up by E:M 9:1 and re-crystallized two times in methanol to yield compound **2** (7.2 mg).

The fraction CAE1 (H:E 9:1; 8.34 g) was taken CP-CC with eluent of H:E (gradient, 0 to 100% E) to give 6 subfractions (CAE1.1-1.6). The subfraction CAE1.4 (H:E 9:1, 1.42 g) was continued to perform CP-CC with H:E 95:5 to get 4 subfractions (CAE1.4.1-1.4.4). The subtraction CAE1.4.2 (H:E 95:5, 0.12 g) was re-crystallized in *n*-hexane to obtain compound **3** (9.2 mg).

### 2.4. Physical characteristic and spectral data

**Oleanolic acid (1):** A white amorphous powder, mp. 306-308°C, lotus purple chromatographic stain, no luminescence under UV lamp. ESI-MS *m/z* 455 [M-H]<sup>-</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 600 MHz,  $\delta_H$  ppm, *J* Hz) and <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 150 MHz,  $\delta_C$  ppm) (Table 4).

**20-hydroxyecdysone (2):** White needle-shaped crystals, mp. 242-244°C, purple pink TLC stain, luminescence under UV lamp. ESI-MS *m/z* 481 [M+H]<sup>+</sup>; <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>, 500 MHz,  $\delta_H$  ppm, *J* Hz) and <sup>13</sup>C-NMR (DMSO-d<sub>6</sub>, 125 MHz,  $\delta_C$  ppm) (Table 4).

**$\beta$ -Spinasterol (3):** White needle-shaped crystals, mp. 164-166°C, red purple TCL stain, no luminescence under UV lamp. ESI-HRMS *m/z* 395.3689 [M-H<sub>2</sub>O+H]<sup>+</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz,  $\delta_H$  ppm, *J* Hz) and <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 125 MHz,  $\delta_C$  ppm) (Table 4).

## 3. RESULTS AND DISCUSSIONS

### 3.1. Bioactivities of extracts

Results of antimicrobial activity, anticancer and antioxidant activities of 4 extracts from *Centrostachys aquatica* were presented in Table 1, 2 and 3, respectively.

**Table 1. Results of antimicrobial activity of extracts from *Centrostachys aquatica***

No.	Code	Initial concentration (µg/mL)	Minimum inhibitory concentration (MIC, µg/mL)				Comments
			Gram (-) Bacteria		Gram (+) Bacteria		
			<i>Escherichia coli</i>	<i>Pseudomonas aeruginosa</i>	<i>Bacillus subtilis</i>	<i>Staphylococcus aureus</i>	
	(-)-Control		(-)	(-)	(-)	(-)	
1	CAAc	400	(-)	(-)	(-)	(-)	Negative
2	CAHe	400	(-)	(-)	100	(-)	Positive (1 strain)
3	CAMe	400	(-)	(-)	200	(-)	Positive (1 strain)
4	CAEt	400	(-)	(-)	200	(-)	Positive (1 strain)

It can be seen from the Table 1 that CAHe, CAMe and CAEt samples exhibited resistance to the test strain of *B. subtilis* with MIC values of 100, 200 and

200 µg/mL, respectively. The CAAc extract was completely negative for 4 tested microorganism strains.

**Table 2. Results of anticancer activity of extracts from *Centrostachys aquatica***

No.	Code	Initial concentration (µg/mL)	Cell survival rate (CS value, %)		IC <sub>50</sub> value (µg/mL)	
			Cell trains		Cell trains	
			Hep-G2	LU-1	Hep-G2	LU-1
	DMSO	-	100	100	-	-
	(+) Control	5	3,14±0,71	1,89±0,60	0,32	0,27
1	CAAc	40	98,88±1,09	88,15±1,64	-	-
2	CAHe	40	63,59±2,04	66,39±0,51	-	-
3	CAMe	40	93,48±2,25	93,38±1,62	-	-
4	CAEt	40	53,36±1,52	40,49±0,48	-	27,66

As the results in Table 2, only the sample CAEt showed inhibitory activity on LU-1 cell line (lung cancer) with IC<sub>50</sub> value of 27.66 µg/mL. The

remaining samples did not show cytotoxic activity of 2 cancer cell lines including Hep-G2 (liver cancer) and LU-1 at the tested concentration.

**Table 3. Results of antioxidant activity of extracts from *Centrostachys aquatica***

No.	Code	Initial concentration (µg/mL)	Scavenging capacity (SC, %)	A half scavenging concentration	Comments
				(SC <sub>50</sub> , µg/mL)	
	(+) Control	50	79,24±1,02	12,02	Positive
	(-) Control	-	0	-	Negative
1	CAAc	200	12,21±2,01	-	Negative
2	CAHe	200	10,98±0,41	-	Negative
3	CAMe	200	11,40±0,66	-	Negative
4	CAEt	200	24,63±1,50	-	Negative

(-) Control: DPPH/EtOH + DMSO.

(+) Control: DPPH/EtOH + ascorbic acid.

Table 3 showed the test samples did not exhibit antioxidant activity on the DPPH system at the test concentrations.

In summary, the bioactivity assay showed that the CAEt extract was the most active of the tested extracts. So this extract was chosen to investigate its chemical composition.

### 3.2. Chemical structure elucidation of isolated compounds

Three isolated compounds had some similar characteristics as being white solids, no producing positive reaction to FeCl<sub>3</sub> reagent; it can be inferred that they do not belong to the group of phenolic substances. Typical signals of protons and carbons

in 1D-NMR showed that they had the patterns of triterpene and sterol backbones.

### 3.2.1. Compound 1

Compound **1** was obtained as a white amorphous powder, its mp. was about 306-308°C. It had lotus purple chromatographic spot and no luminescence under UV lamp.

The molecular formula of compound **1** was speculated to be  $C_{30}H_{48}O_3$  (456 amu, seven degrees of unsaturation) on the basis of ESI-MS ( $m/z$  455  $[M-H]^-$ ).

The  $^1H$ -NMR spectrum of compound **1** revealed 7 singlet signals of methyl protons at  $\delta_H$  [0.76 (3H, *s*), 0.77 (3H, *s*), 0.90 (3H, *s*), 0.91 (3H, *s*), 0.93 (3H, *s*), 0.99 (3H, *s*) and 1.13 (3H, *s*)]; one oxygenated methine proton at  $\delta_H$  3.22 (1H, *dd*, 11.4, 4.2); one double-bonded methine proton at  $\delta_H$  5.28 (1H, *t*, 3.6) and about 25 other protons of methine, methylene, hydroxyl groups (Table 4).

The  $^{13}C$ -NMR and DEPT spectra of compound **1** appeared signals of total 30 carbons including 7 methyl, 10 methylene, 5 methine and 8 quaternary

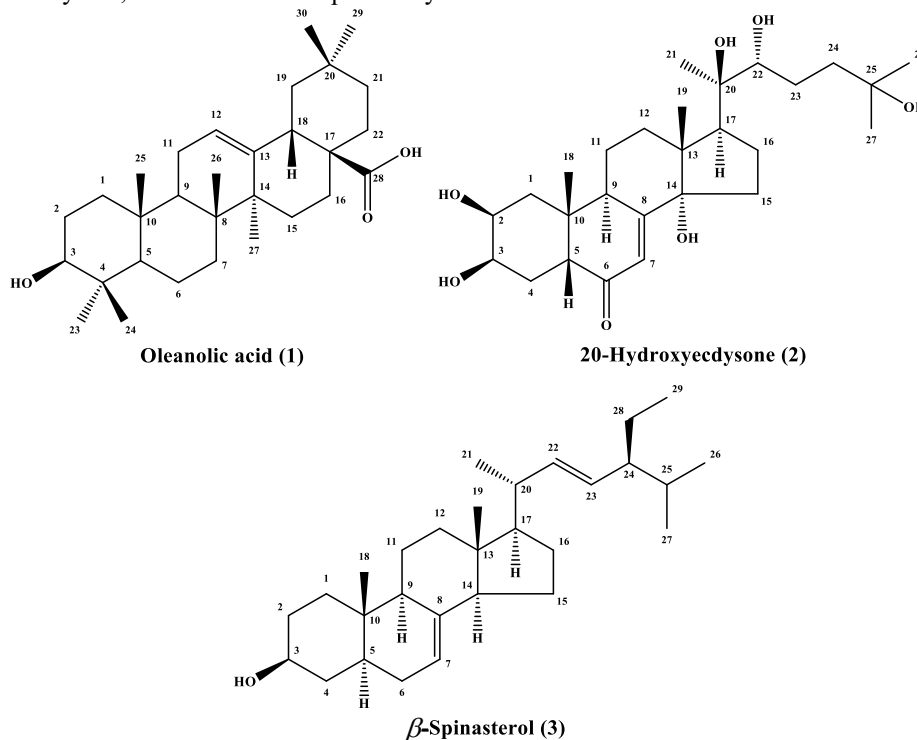
carbons. In which, 2 carbons at  $\delta_C$  122.7 and 143.6 exhibited the presence of a double bond; carbon at  $\delta_C$  182.8 allowed to predict having a carboxylic group; and carbon at  $\delta_C$  79.1 was oxygenated methine group (Table 4).

From mentioned 1D-NMR data, compound **1** gave the characteristic spectra pattern of a pentacyclic triterpene. Based on the spectral data analysis, comparison with those given in the literature (Zuhail et al., 2009) and checking with its own 2D-NMR spectra, compound **1** was identified as oleanolic acid (Figure 2).

Oleanolic acid could be used to prevent the majority of the most common diseases of civilization *i.e.* cancer, cardiovascular diseases, atherosclerosis or diabetes (Paszal-Jaworska et al., 2014).

### 3.2.2. Compound 2

Compound **2** was isolated as white needle-shaped crystals, its mp. was about 242-244°C. It had purple pink TLC stain and luminescence under UV lamp, that proved there was a conjugate system in its chemical structure.



**Figure 2. Chemical structures of isolated compounds**

The molecular formula of compound **2** was established as  $C_{27}H_{44}O_7$  (480 amu, six degrees of unsaturation) by ESI-MS ( $m/z$  481.1  $[M+H]^+$ ).

$^1H$ -NMR spectrum of compound **2** appeared typical proton signals of 5 methyl groups at  $\delta_H$  [0.76 (3H, *s*), 0.84 (3H, *s*), 1.06 (6H, *s*), and 1.08 (3H, *s*)]; 3 oxygenated methine proton at  $\delta_H$  [3.12 (1H, *d*, 5.0),

3.56 (1H, s), and 3.77 (1H, s)]; one double-bonded methine proton at  $\delta_H$  5.63 (1H, s) and about 25 other protons of methine, methylene, hydroxyl groups (Table 4).

$^{13}\text{C}$ -NMR and DEPT spectra exhibited signals of total 27 carbons containing 5 methyl, 8 methylene, 7 methine and 7 quaternary carbons. There were figurative signals as one ketone carbon at  $\delta_C$  202.6; 6 oxygenated methine and quaternary carbons at  $\delta_C$  [66.6, 66.7, 68.7, 75.7, 76.2 and 82.9]; 5 methyl carbons at  $\delta_C$  [17.1, 20.9, 23.8, 29.0 and 29.9]. In addition, methine carbon at  $\delta_C$  120.4 and quaternary carbon at  $\delta_C$  165.2 confirmed the presence of a C=C double bond (Table 4).

With the above analyzed spectral characteristics, it is possible to predict compound **2** belonging to the steroid group. The 1D-NMR spectral data of compound **2** were similar to those of 20-

hydroxyecdysone (Figure 2) given in the literature (Vokac et al., 1998). Moreover, all correlation signals between protons and carbons in HSQC and HMBC spectral data of compound **2** conformed with the mentioned chemical structure, so compound **2** was determined to be 20-hydroxyecdysone.

A special available activity of 20-hydroxyecdysone was reducing moult cycle duration of the edible freshwater crab *Travancoriana schirnerae* (Raghavan and Ayanath, 2019).

### 3.2.3. Compound 3

Compound **3** was also received as white needle-shaped crystals, its mp. was 164-166°C. It had red purple TCL spot and no luminescence under UV light.

**Table 4. 1D-NMR spectral data of isolated compounds**

C-position	Compound 1		Compound 2		Compound 3	
	$^1\text{H}$	$^{13}\text{C}$ $^1\text{H}$	$^{13}\text{C}$ $^1\text{H}$	$^{13}\text{C}$ $^1\text{H}$	$^{13}\text{C}$	
1	1.60, 0.95	38.4	1.59, 1.29	36.6	1.09, 1.82	37.2
2	1.75, 1.56	27.2	3.56 (1H, s)	66.7	1.39, 1.77	31.5
3	3.22 (1H, dd, 11.4, 4.2)	79.1	3.77 (1H, s)	66.6	3.56-3.62 (1H, m)	71.1
4		38.8	1.57, 1.48	31.5	1.27, 1.70	38.0
5	0.73 (1H, s)	55.2	2.20 (1H, dd, 13.0, 30.0)	50.1	1.40	40.3
6	1.52, 1.33	18.3		202.6	1.22, 1.74	29.7
7	1.43, 1.29	32.6	5.63 (1H, s)	120.4	5.15 (1H, brs)	117.5
8		39.3		165.2		139.6
9	1.54	47.6	3.01 (1H, s)	33.1	1.65	49.5
10		37.1		37.6		34.2
11	1.89, 1.87	23.4	1.86, 1.64	20.0	1.48	21.6
12	5.28 (1H, t, 3.6)	122.7	1.79, 1.52	30.3	1.99, 2.02	39.5
13		143.6		46.8		43.3
14		41.6		82.9	1.81	55.1
15	1.13, 1.06	27.7	2.01 (1H, d, 8.5), 1.72	30.8	1.40, 1.52	23.0
16	1.95, 1.99	23.0	1.62 (2H)	20.2	1.25	28.5
17		46.5	2.26 (1H, t, 8.5)	48.7	1.25	55.9
18	2.82 (1H, dd, 9.6, 4.2)	41.0	0.84 (3H, s)	23.8	0.81 (3H, s)	13.0
19	1.64, 1.18	45.9	0.76 (3H, s)	17.1	0.55 (3H, s)	12.1
20		30.7		75.7	2.04	40.8
21	1.38, 1.24	33.8	1.06 (3H, s)	20.9	1.03 (3H, d, 6.5)	21.4
22	1.78, 1.55	32.4	3.12 (1H, d, 5.0)	76.2	5.17 (1H, dd, 15.0, 8.5)	138.2
23	0.99 (3H, s)	28.1	1.50, 1.12	26.1	5.03 (1H, dd, 15.0, 8.5)	129.5
24	0.77 (3H, s)	15.6	1.66, 1.24	41.4	1.55	51.3
25	0.91 (3H, s)	15.3		68.7	1.55	31.9
26	0.76 (3H, s)	17.1	1.08 (3H, s)	29.9	0.85 (3H, d, 6.5)	21.1
27	1.13 (3H, s)	25.9	1.06 (3H, s)	29.0	0.82 (3H, s)	19.0
28		182.8			1.18, 1.42	25.4
29	0.90 (3H, s)	33.1			0.80 (3H, t, 2.0)	12.2
30	0.93 (3H, s)	23.6				

Note: Compounds **1** and **3** were recorded in  $\text{CDCl}_3$ ; compound **2** was recorded in  $\text{DMSO}-d_6$ .

The molecular formula of compound **3** was speculated to be C<sub>29</sub>H<sub>48</sub>O (412 amu, six degrees of unsaturation) on the basis of ESI-HRMS *m/z* 395.3689 [M-H<sub>2</sub>O+H]<sup>+</sup>.

Most of 1D-NMR spectral signals of compound **3** were similar to those of compound **2** (Table 4). However, compound **3** had two carbons (one methyl group) more than compound **2**. There were only one oxygenated methine group at  $\delta_C$  71.1 and with it the disappearance of the carbonyl group (Table 4).

Spectral data of compound **3** were compared with those given in the literature (Ragasa and Lim, 2005), 2D-NMR spectra of **3** were also used to check the fit of the predicted structure. As a result, compound **3** was identified as  $\beta$ -spinasterol (Figure 2).

$\beta$ -Spinasterol was well evaluated its antiproliferative activity against human cancer cell lines HeLa and murine cancer cell line RAW 264.7 (Meneses-Sagrero et al., 2017).

#### 4. CONCLUSIONS

Extracts from the whole plant of *Centrostachys aquatica* collected in Can Tho city were tested

bioactivity towards human applications and studied chemical components for the first time.

The CAEt extract exhibited resistance to the test strain of *B. subtilis*, inhibited for LU-1 cell line. CAHe and CAME extracts showed resistance to the test strain of *B. subtilis*. Other above-declared experiments were negative.

Oleanolic acid, 20-hydroxyecdysone and  $\beta$ -spinasterol were isolated for the first time from CAEt extract of *Centrostachys aquatica*.

The interesting result was ability to inhibit human lung cancer cell of CAEt extract. Therefore, it is necessary to further isolate and test the anticancer activity of the purified compounds from this extract in order to be used as a medicinal plant. This research is still going on; the next results will be published as soon as possible.

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